

# Ambient Ammonia Concentrations Across New York State

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# Acknowledgments

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# Objectives

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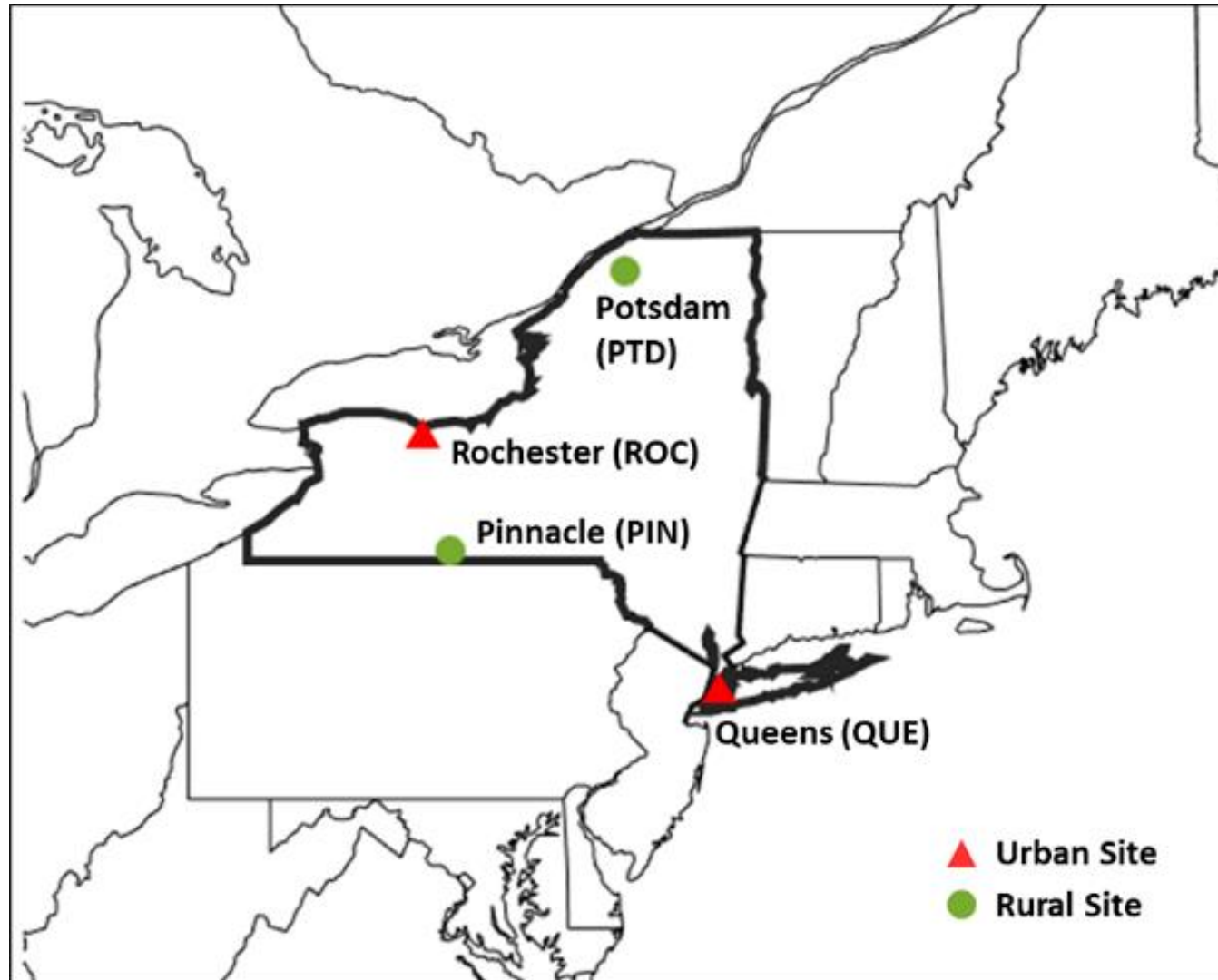
- The objectives of this project were to measure ambient  $\text{NH}_3$  in four locations across New York to identify its spatial and temporal distributions, source types, and its role in secondary particle formation.
- The four sites included 2 urban sites (Queens and Rochester, NY) and 2 rural sites (Pinnacle State Park and Potsdam, NY).
- At each site, AMoN passive samplers were also deployed so we could compare the active monitor with the integrated passive samples.

# Manuscript

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This work has been written up in a manuscript currently under review for the Journal of Geophysical Research - Atmospheres

# Map of the four $\text{NH}_3$ monitoring sites.

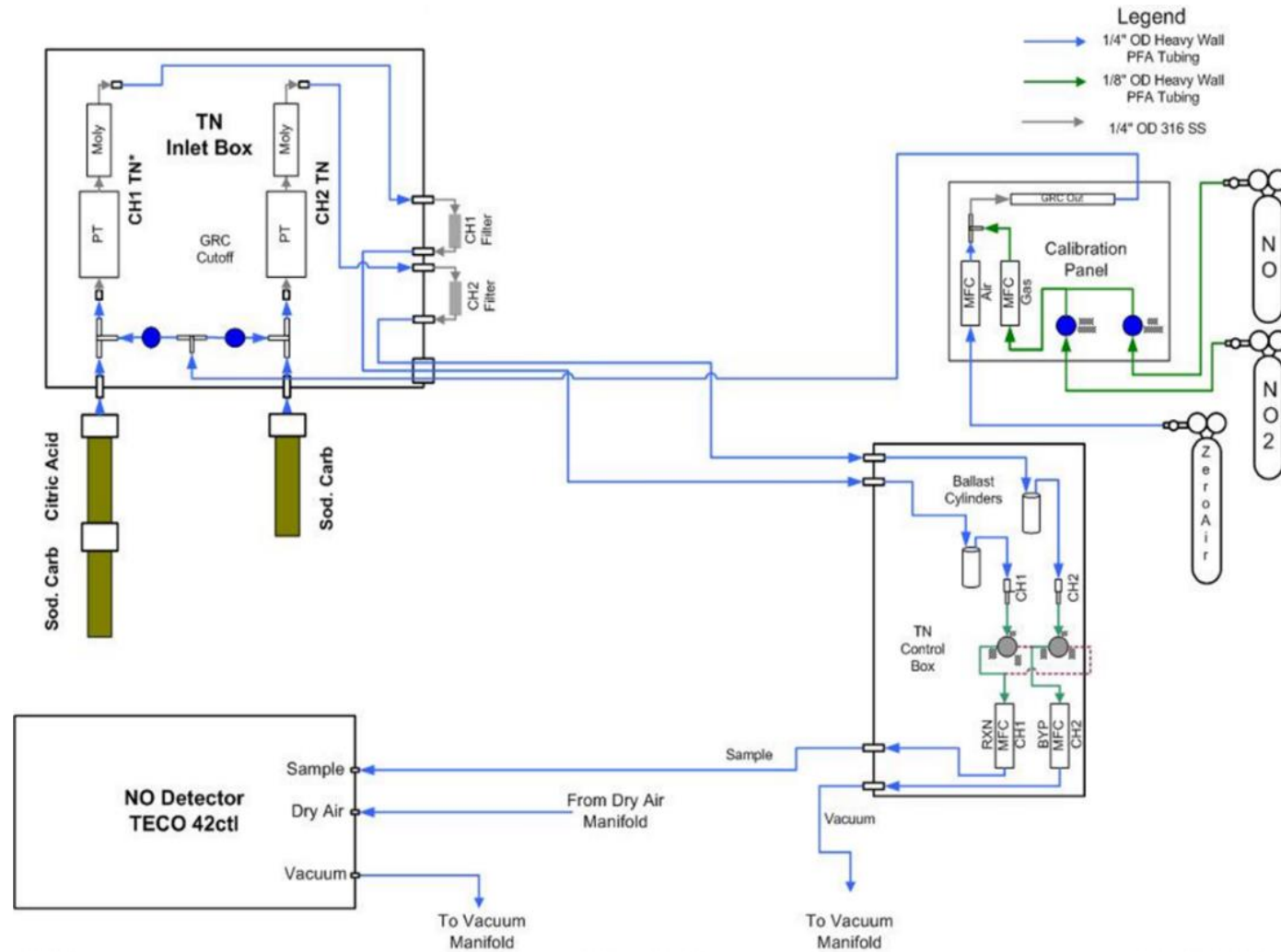


# Active Monitor

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- The active system used a denuder difference technique through a dual-channel nitric oxide-ozone (NO-O<sub>3</sub>) chemiluminescence detector (CLD) system with 5-minute time resolution for data reporting and analysis (Edgerton et al., 2007).
- The flow in channel 1 was directed through a 600°C Pt mesh converter to oxidize the reduced nitrogen species (primarily NH<sub>3</sub> and particulate NH<sub>4</sub><sup>+</sup>) to NO<sub>y</sub> followed by a 350°C Mo converter to reduce the NO<sub>y</sub> to NO.
- The flow in channel 2 was directed to a citric acid-coated annular denuder to remove all NH<sub>3</sub> before being directed to the Pt and Mo converters.
- The NH<sub>3</sub> concentrations were defined as the differences between the nitrogenous species measured in the two channels.

# Active Monitor



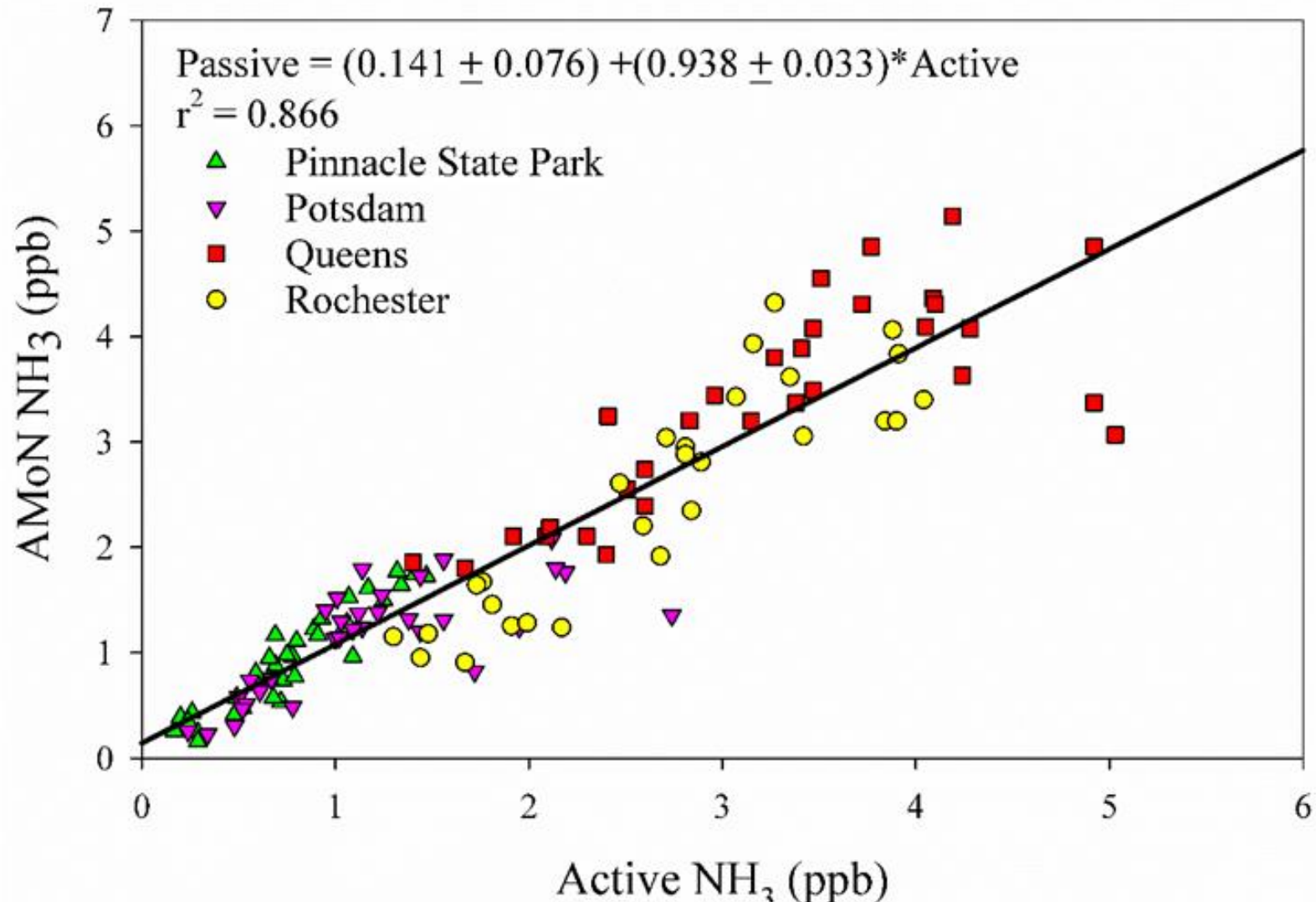
# Passive Monitor

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- Passive diffusion samplers were Radiello from Sigma-Aldrich.
- $\text{NH}_3$  was sorbed onto the interior surface of samplers based on diffusion, and the  $\text{NH}_3$  samples were analyzed and quality controlled by the NADP under AMoN
- Ammonium ion is quantified by visible spectrometry as indophenol:
  - at basic buffered pH ammonium ion reacts with phenol and sodium hypochlorite, with pentacyanonitrosylferrate catalysis (in the following cyanoferrate), to form indophenol.
  - The reaction product is intensely colored in blue, and its absorbance measured at 635 nm.



# Comparison of Active and Passive Monitors



# Comparison of Active and Passive Monitors

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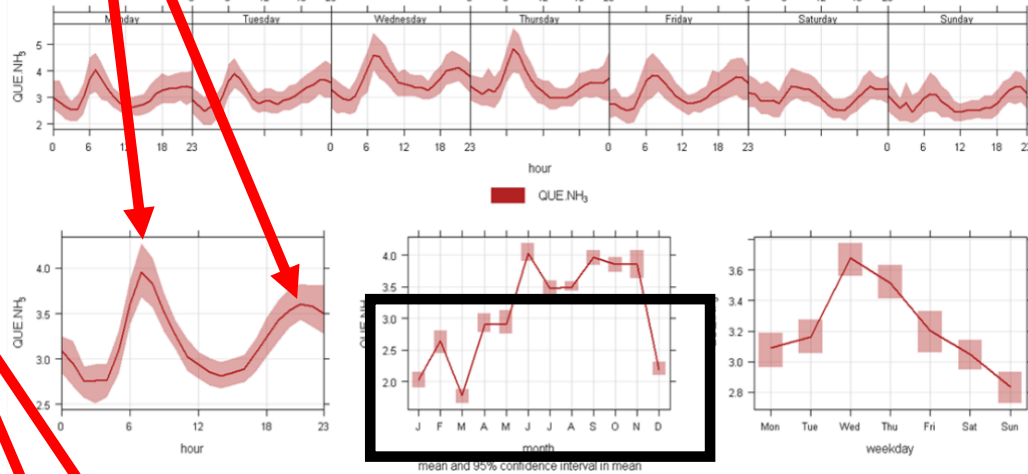
- There is good agreement between the passive and active monitors.
- Agreement was similar across rural and urban sites
- Active monitors values were about 6% larger than the passive monitors, but agreement is within the uncertainties in both values.

Rush Hour Peaks

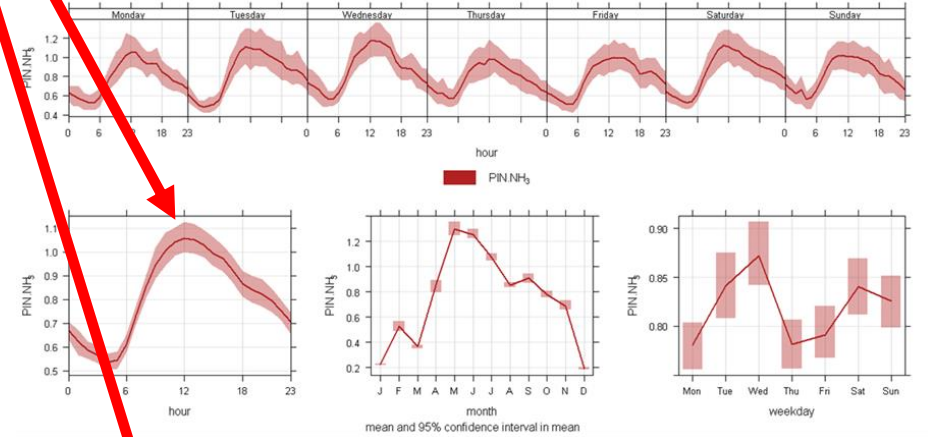
Midafternoon Peaks

# Temporal Variability

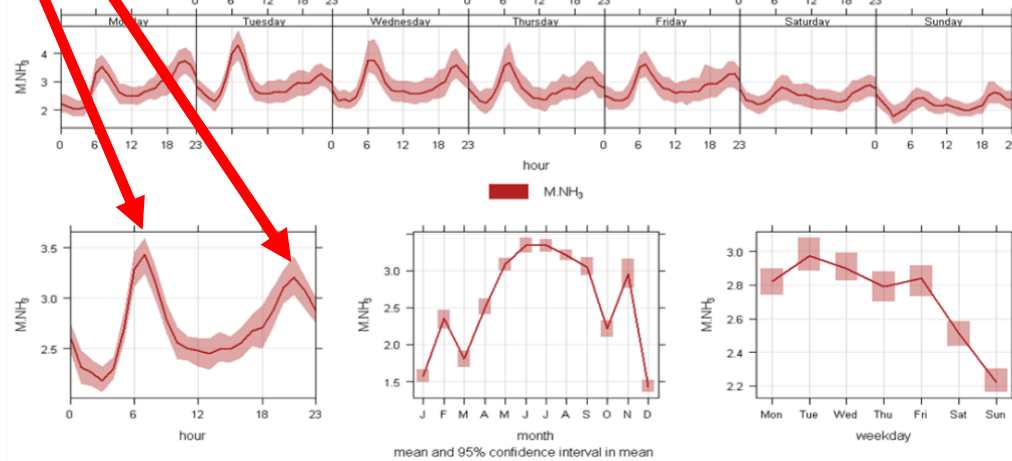
Queens



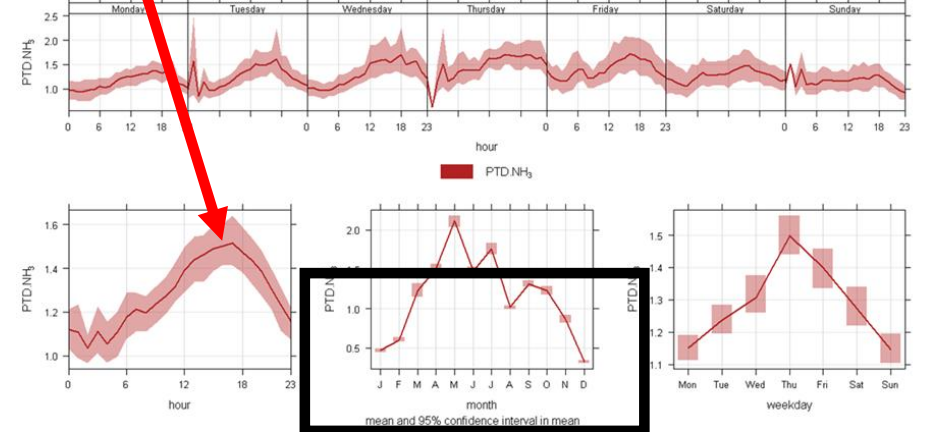
Pinnacle State Park



Rochester



Potsdam



# Mean Values

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- The mean ( $\pm$  SD)  $\text{NH}_3$  concentrations measured using active systems in QUE, ROC, PTD, and PIN were  $3.22 \pm 2.23$ ,  $2.84 \pm 1.91$ ,  $1.29 \pm 1.12$ , and  $0.82 \pm 0.64$  ppb, respectively.
- $\text{NH}_3$  concentrations at the urban sites (QUE and ROC) were significantly higher (Kruskal-Wallis test,  $p < 0.01$ ) than that at the rural sites (PTD and PIN).

# Inventory Emissions

Table 1. Local NH<sub>3</sub> emissions and emissions densities (county-based) for the four sites based on NEI 2014.

Monitoring site	Counties	Total NH <sub>3</sub> emission (ton/year)	Total NH <sub>3</sub> emission density (ton/year/km <sup>2</sup> )	Non-agricultural emission (%)
NYS total		33337	0.24	63
QUE	Queens	1101	2.38	99
ROC	Monroe	748	0.21	58
PTD	St. Lawrence	1027	0.14	1
PIN	Steuben	960	0.26	1

# Inventory Emissions

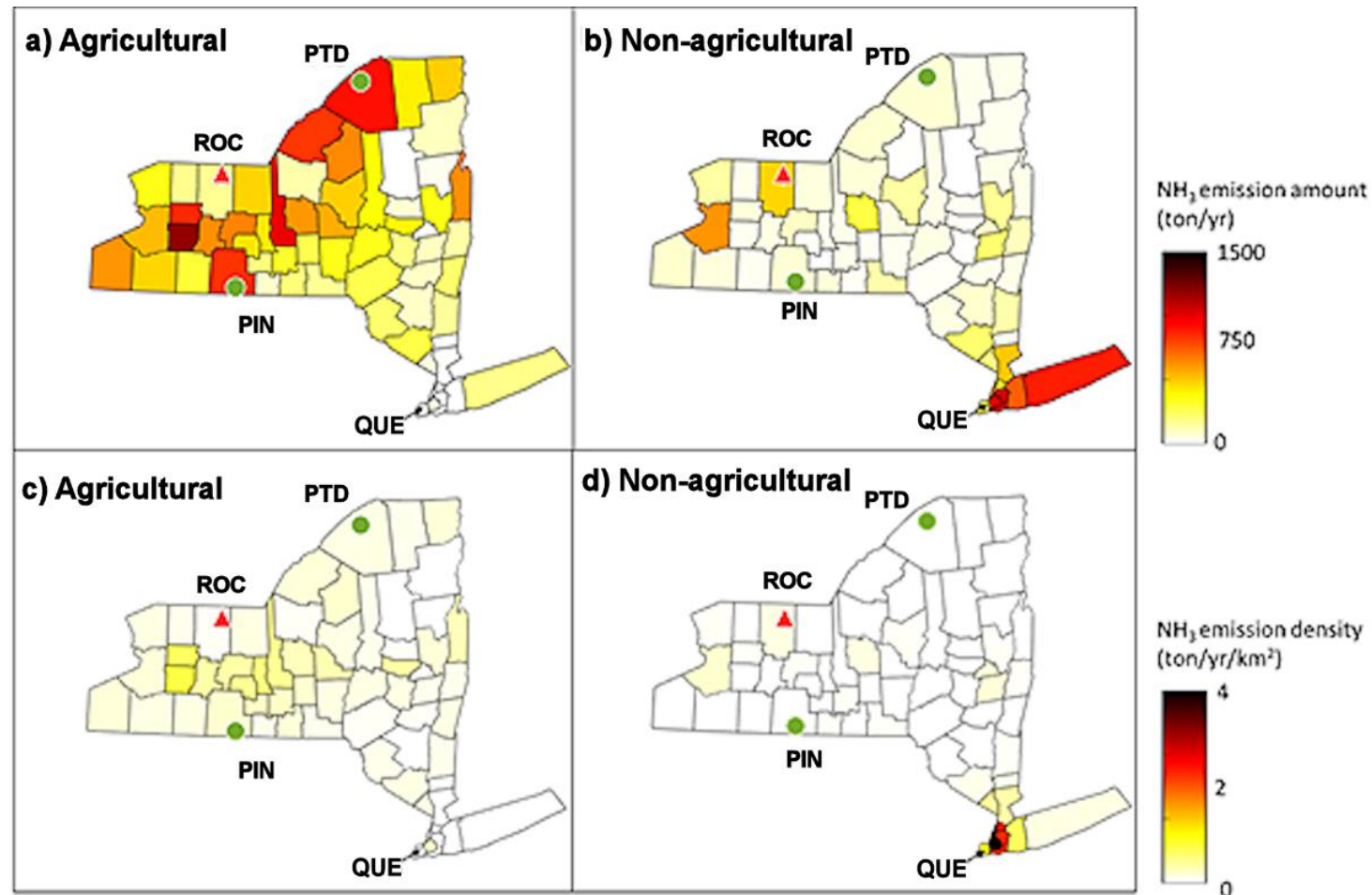


Figure S7.  $\text{NH}_3$  emissions rates and density of agricultural and non-agricultural sources in each NYS county.

# Inventory Emissions

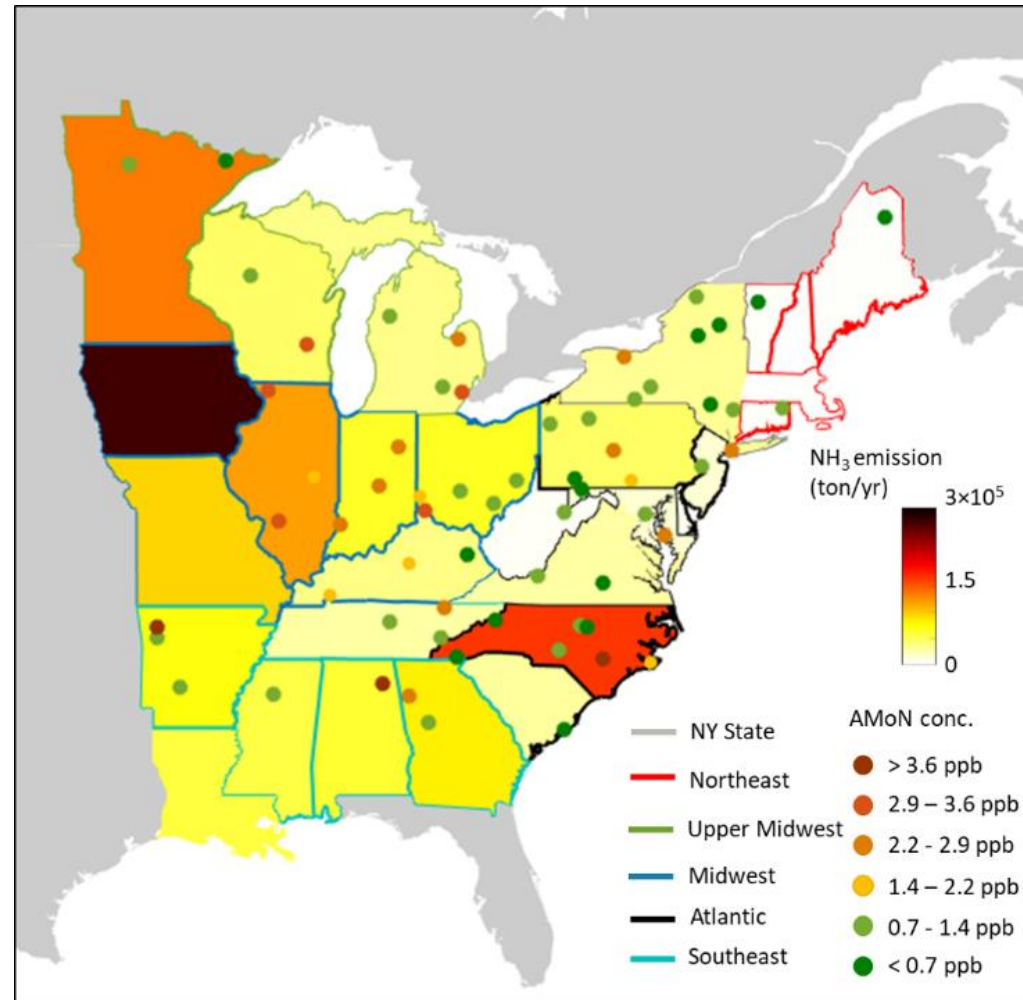


Figure 6. Annual  $\text{NH}_3$  emissions in the central and eastern U.S. (divided into five regions with different border color) and AMoN site (presented as circles) concentrations.

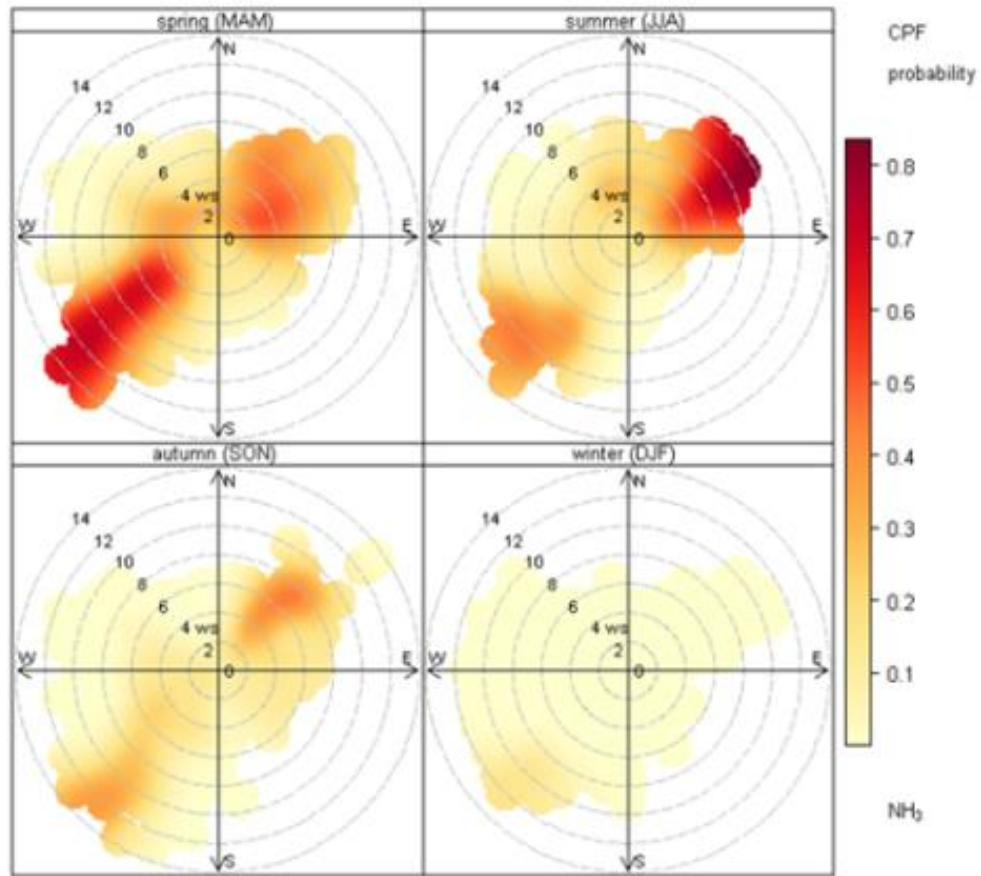
# Proportions of Emissions from Source Areas

Table 2. Non-negative regression results for the NH<sub>3</sub> concentrations at the four sites.

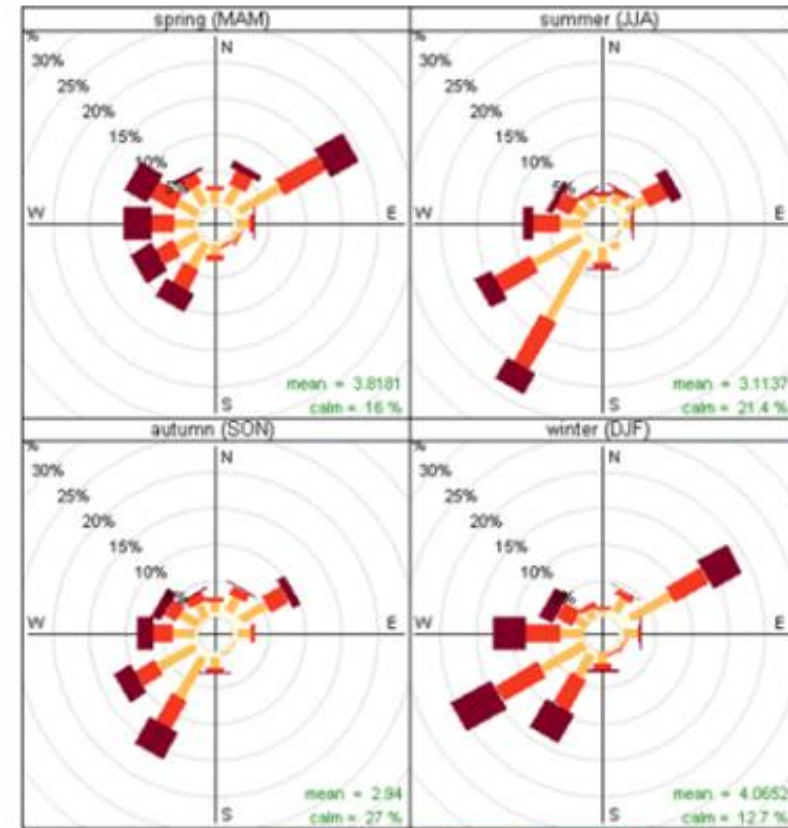
	Midwest	Upper Midwest	Northeast	Atlantic	Southeast	Local	R <sup>2</sup>
PIN	0.00033	0.0138	0.291	0.00	0.00	0.00	0.76
PTD	0.00	0.0196	0.00	0.0296	0.00	0.00	0.60
ROC	0.00	0.00	0.837	0.00	0.00	1.31	0.42
QUE	0.00	0.00	0.703	0.00	0.00	2.00	0.30



# Local Emission Directions - Potsdam

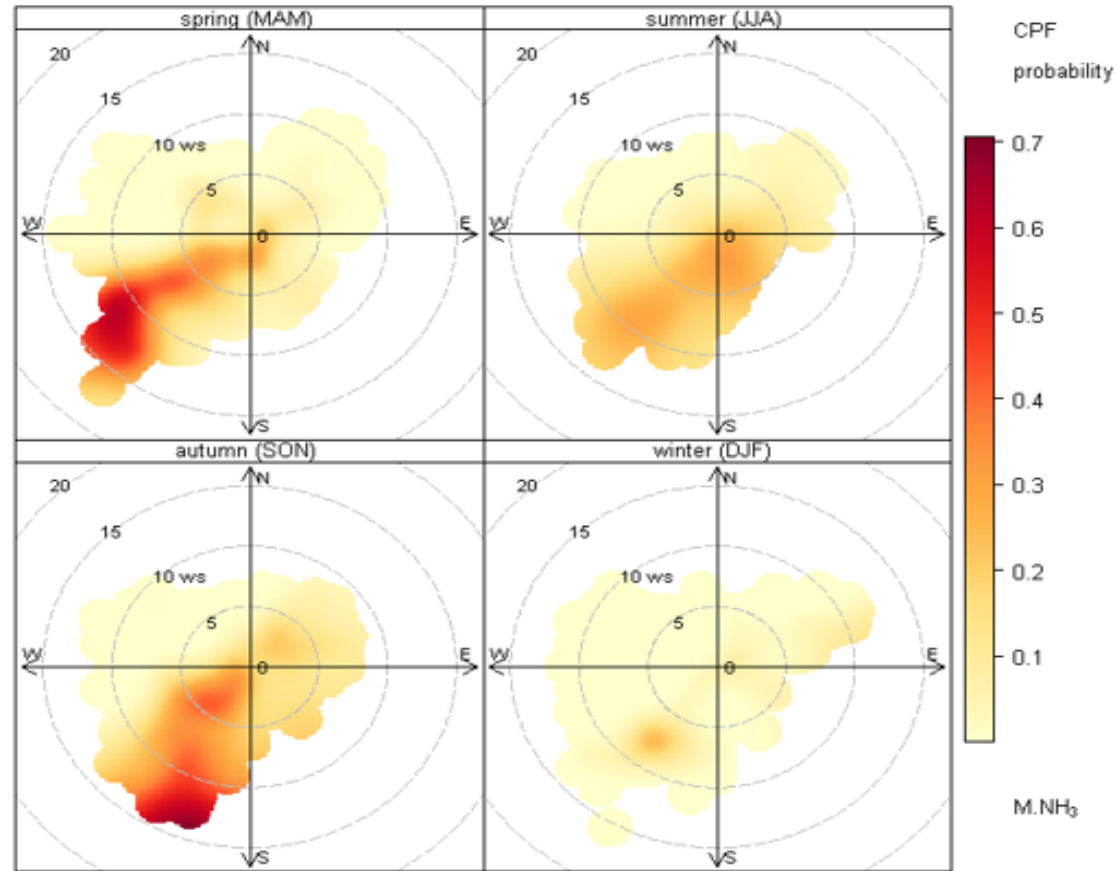


CPF at the 75th percentile ( $\approx 1.7$ )

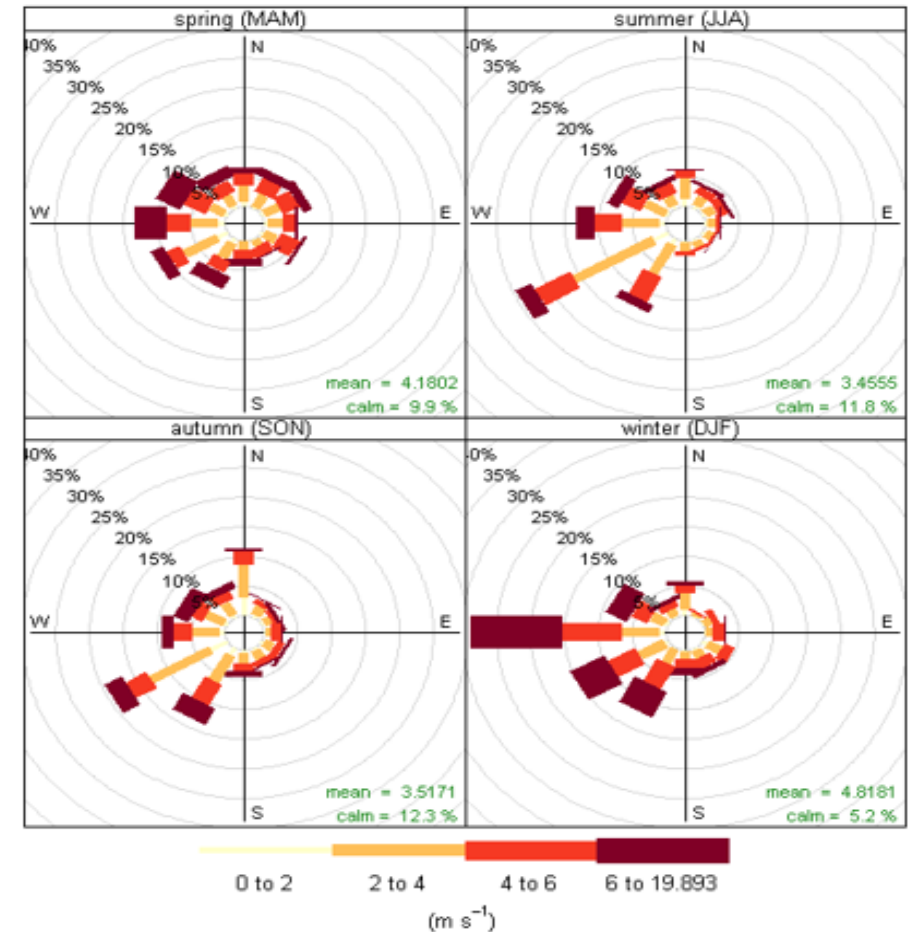


Frequency of counts by wind direction (%)

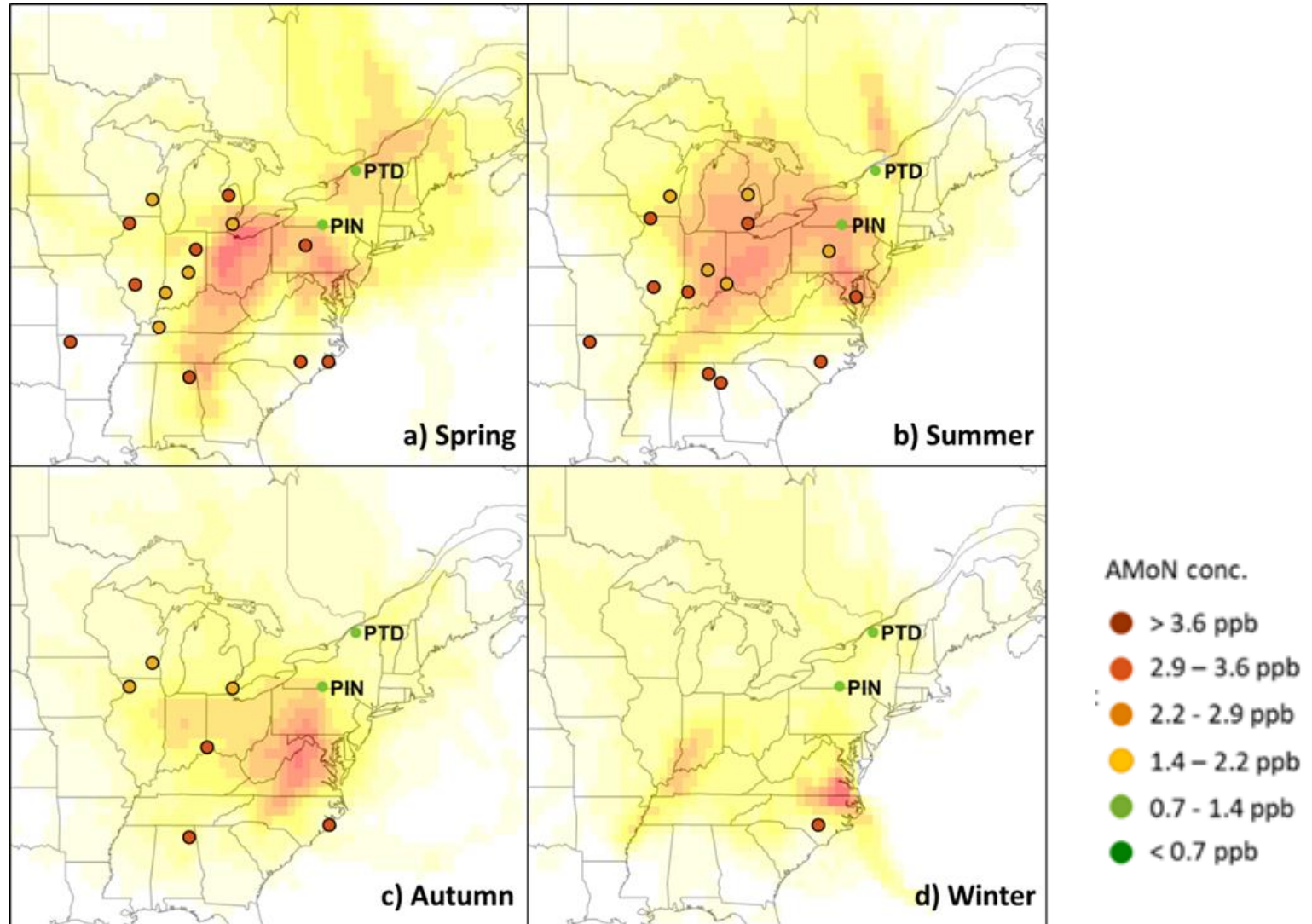
# Local Emission Directions - Rochester



CPF at the 75th percentile (=3.9)



# Distant Emissions - Rural



# Conclusions

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- Active and passive monitors were comparable; Active system measured 6.2% higher  $\text{NH}_3$  concentrations
- Urban sites (QUE and ROC) had higher  $\text{NH}_3$  concentrations than the rural sites (PIN and PDM) emphasizing the importance of local sources such as traffic and population centers.
- Diel variations at the urban sites coincided with typical traffic patterns, while at the rural sites, concentrations tracked ambient temperature changes.
- Seasonal patterns were weaker at the urban sites and appeared to depend more on changes in the dispersion conditions. Rural sites show low winter values when soil activity is low.
- Urban  $\text{NH}_3$  emissions, mainly from vehicles and stationary SCRs, have impacts on urban  $\text{NH}_3$  concentrations and resulting urban particulate air pollution (nitrate and sulfate) even though agricultural  $\text{NH}_3$  emissions nominally dominated the reported total NYS  $\text{NH}_3$  emissions.
- The ammonia emission factor for residential combustion of natural gas substantially overestimates those emissions. Emission factor needs to be updated given the changes in the efficiency of these heating appliances.

Thank you for your attention!

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Questions?